## TOTAL SYNTHESIS OF THE (+)-PHEROMONE OF THE MALE SWIFT MOTH HEPIALUS HECTA L.

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ABSTRACT The (+)-enantiomer of the pheromone of the male swift moth has been synthesized in 6 steps with an overall yield of 16%.

The isolation and structure determination of the pheromones (1 and 2) of the male swift moth <u>Hepialus hecta L</u>. were reported by Francke et al. in 1985.<sup>2</sup> The structures of 1 and 2 were deduced from mass spectral fragmentation patterns and analysis of the  $^{1}$ H/ $^{13}$ C NMR spectra. However, the small quantities of material available from natural sources (ca. 200  $\mu$ g of <u>1</u> and <u>2</u>, respectively) precluded determination of the absolute stereochemistry of the pheromones.



As part of our research program for the total synthesis of the dienoyl tetramic acid antibiotics tirandamycin A/B and streptolydigin, we had previously developed a general approach to the synthesis of bicyclic ketal moieties similar to  $\underline{1}$  and  $\underline{2}$ .<sup>3</sup> In this Communication, we report a six step synthesis of the (+)-enantiomer of <u>l</u> which confirms the structure of the natural product and establishes the absolute stereochemistry for this enantiomer as 1R, 3S, 5R.

Reduction of  $\beta$ -keto ester 3 with baker's yeast<sup>4</sup> gave S-alcohol  $4^5$  in 75% isolated yield and >95% enantiomeric excess. This reduction can be routinely

performed on 10-20 gram quantities of the ester and facilitated preparation of multigram quantities of synthetic intermediates.

As outlined in the Scheme, protection of the alcohol function as the TBDMS silyl ether<sup>6</sup> under standard conditions (93%) and reduction of the ester with DIBAL<sup>7</sup> produced §-aldehyde 5<sup>8</sup> in 82% yield. Condensation of aldehyde 5 with 5-lithio-2,3-dimethylfuran<sup>3</sup> (Et<sub>2</sub>0, -78°C) gave a 1:1 mixture of furan-alcohols 6<sup>9</sup> and  $\underline{7}^{10,11}$  (79%). The stereochemistry of the sidechain oxygen functionalities in 6 and 7 were assigned by analogy with related substances<sup>3</sup> and was confirmed by subsequent transformations (<u>vide infra</u>).

Treatment of <u>anti</u>-alcohol <u>6</u> with <u>m</u>CPBA in CH<sub>2</sub>Cl<sub>2</sub><sup>3</sup> gave a mixture of pyranones (91%) which were cyclized to  $(+)-\underline{1}^{12}$  by treatment with dilute HF in aqueous acetonitrile (78%). The spectral data and GC characteristics of synthetic  $(+)-\underline{1}$  were identical to the reported spectral data<sup>2,13</sup> and thus confirmed the original structure assignment. Using this six step synthesis,  $(+)-\underline{1}$  was prepared in an overall yield of 16% from readily available starting materials.

Scheme



Oxidation of alcohol  $\underline{7}$  with mCPBA followed by HF treatment produced lactone  $\underline{8}^{14}$  by rearrangement of the intermediate pyranone mixture. The stereochemistry of  $\underline{8}$  was assigned by analysis of the <sup>1</sup>H NMR spectrum which was similar to the <sup>1</sup>H NMR spectrum of lactone  $\underline{9}$ .<sup>3</sup> The structure of lactone  $\underline{9}$  had been determined previously by single crystal X-ray analysis.<sup>15</sup>



Thioketalization of <u>1</u> (propanedithiol, BF<sub>3</sub> •Rt<sub>2</sub>O) gave dithiane <u>10<sup>16</sup></u> which upon Raney nickel desulfurization furnished alkene <u>2</u>, albeit in low yield. We are presently engaged in improving the yield of the desulfurization procedure.



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## References and Notes

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- DeShong, P.; Ramesh, S.; Elango, V.; Perez, J. J. J. <u>J. Am. Chem. Soc.</u> 1985, <u>107</u>, 5219, and references cited therein.
- 4. Sih, C. J.; Zhou, B. N.; Gopalan, A. S.; Shieh, W. R.; Van Middlesworth, F. in W. Bartmann, B. Trost: Selectivity-A Goal for Synthetic Efficiency, Workshop Conferences Hoechst Vol. 14, Verlag Chemie, Weinheim 1983, p. 251, and references cited therein.

- 5. <u>S</u>-Alcohol <u>4</u>:  $[\alpha]_D^{20}$  =+21.9<sup>•</sup> (C=2.7, CHCl<sub>3</sub>),  $[\alpha]^{23}$  (Lit.<sup>4</sup>) =+21.7<sup>•</sup> (C=5.9, CHCl<sub>3</sub>); IR (CCl<sub>4</sub>) 3540, 1720 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) & 0.88 (t, 3, J=7), 0.96 (t, 3, J=7), 1.15-1.71 (m, 14), 2.39 (dd, 1, J=16, 9), 2.52 (dd, 1, J=16, 4), 3.87-3.99 (m, 1), 4.10 (t, 2, J=7).
- 6. Corey, E. J.; Venkateswarlu, A. <u>J. Am. Chem. Soc.</u>, 1972, <u>94</u>, 6190.
- 7. Winterfeldt, E. Synthesis, 1975, 617.
- 8. <u>S</u>-Aldehyde <u>5</u>:  $[\alpha]_D^{20} = +2.9^{\circ}$  (C=2.39, CHCl<sub>3</sub>); IR (CCl<sub>4</sub>) 2710, 1725 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) & 0.06 (s, 3), 0.07 (s, 3), 0.87 (s, 9), 0.90 (t, 3, J=7), 1.49-1.63 (m, 2), 2.51 (dd, 2, J=6, 2), 4.08-4.19 (m, 1), 9.81 (t, 1, J=2).
- 9. Alcohol <u>6</u>:  $[\alpha]_D^{20}$ =+24.7 (C=2.47, CHCl<sub>3</sub>); IR (CCl<sub>4</sub>): 3610 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) & 0.10 (s, 3), 0.12 (s, 3), 0.90 (t, 3, J=7), 0.91 (s, 9), 1.56-1.70 (m, 2), 1.91 (s, 3), 1.80-2.14 (m, 2), 2.19 (s, 3), 3.29 (d, 1, J=3), 3.90-3.98 (m, 1), 4.89 (dt, 1, J=10, 3), 6.00 (s, 1).
- 10. Alcohol <u>7</u>: [α]<sup>20</sup>=+10.6° (C=2.55, CHCl<sub>2</sub>); IR (CCl<sub>4</sub>) 3610 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.09 (s, 3), 0.11 (s, 3), 0.89 (dd, 3, J=10, 7), 0.91 (s, 9), 1.49-1.63 (m, 2), 1.91 (s, 3), 1.94-1.98 (m, 2), 2.19 (s, 3), 3.25 (d, 1, J=2), 3.83-3.89 (m, 1), 4.76 (td, 1, J=6, 2), 6.01 (s, 1).
- 11. Alcohol <u>7</u> could be converted into <u>anti</u>-alcohol <u>6</u> by oxidation with BaMnO<sub>4</sub> (80%)<sup>3</sup> followed by reduction with Zn(BH<sub>4</sub>)<sub>2</sub> (90%) to give a 1:1 mixture of <u>6</u> and <u>7</u>. Attempted conversion of <u>7</u> to alcohol <u>6</u> by inversion under Mitsunobulike conditions led to extensive decomposition of the starting alcohol.
- Bnone 1: [α]<sup>20</sup><sub>D</sub> =+346.4° (C=1.94, CHCl<sub>3</sub>); IR (CCl<sub>4</sub>) 1685, 1625 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.87 (t, 3, J=7), 1.52 (s, 3), 1.37-1.91 (m, 4), 1.95 (d, 3, J=1), 3.56-3.67 (m, 1), 4.31 (dt, 1, J=6, 1), 6.15 (s, 1). <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 9.1, 19.3, 24.7, 29.0, 32.1, 69.2, 75.0, 95.9, 126.8, 156.8, 197.1. Exact mass calcd. for C<sub>11</sub>H<sub>18</sub>O<sub>3</sub>: 196.1099. Found: 196.1100.
- 13. Professor W. Francke, Institut für Organische Chemie, Hamburg, has confirmed that a sample of (+)-1 prepared by synthesis was spectroscopically and chromatographically identical with the natural material.
- 14. Lactone 8: IR (CC14) 3600, 1740 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDC1<sub>3</sub>) & 1.02 (t, 3, J=7), 1.29 (d, 3, J=7), 1.48 (s, 3), 1.61-1.81 (m, 3), 2.20-2.32 (m, 2), 2.90 (t, 1, J=10), 4.03-4.15 (m, 1), 4.56 (ddd, 1, J=11, 10, 7).
- 15. The structure of lactone <u>9</u><sup>3</sup> has been determined by single crystal X-ray analysis; P. DeShong, M. Parvez, M.-T. Lin, unpublished results. The <sup>1</sup>H NMR spectrum of <u>9</u> (360 MHz) displays J<sub>3</sub>, 4=10 Hz and J<sub>3</sub>, 9=10 Hz, respectively. The <sup>1</sup>H NMR spectrum of <u>8</u> has J<sub>3</sub>, 4=10 Hz and J<sub>3</sub>, 9=10 Hz.
- 16. Dithiane <u>10</u>: <sup>1</sup>H NMR (CDCl<sub>3</sub>) & 0.90 (t, 3, J=7), 1.26-1.62 (m, 2), 1.46 (s, 3), 1.69 (d, 3, J=1.5), 1.73-2.12 (m, 3), 2.23 (ddd, 1, J=14, 3, 1), 2.69-2.81 (m, 3), 2.98 (ddd, 1, J=14, 11, 4), 3.55-3.69 (m, 1), 4.52 (d, 1, J=6), 5.77 (t, 1, J=1.5). Exact mass calcd. for C<sub>14</sub>H<sub>22</sub>O<sub>2</sub>S<sub>2</sub>: 286.1061. Found: 286.1056.

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